# Basics of QM I. 

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〔SZÉCHENYI
EGYETEM

Fizika és Kémia
Tanszék

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\hat{H} \Psi(\underline{r}, t)=i \hbar \frac{\partial \Psi(\underline{r}, t)}{\partial t}
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$\hat{\boldsymbol{H}}$ - Hamilton-operator of the system
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$\hbar=\frac{\boldsymbol{h}}{2 \pi}$ - modified Planck-constant $\left(\boldsymbol{h}=6,63 \cdot 10^{-34} \mathrm{Js}\right)$

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Fundamental assignments:

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Let's see how the Hamilton-operator of a mass point is defined!

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Let's move on to the Hamilton operator:

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\begin{gathered}
\hat{\boldsymbol{H}}=\frac{\mathbf{1}}{2 \boldsymbol{m}}\left(\hat{\boldsymbol{p}}_{x} \hat{\boldsymbol{p}}_{x}+\hat{\boldsymbol{p}}_{y} \hat{\boldsymbol{p}}_{y}+\hat{\boldsymbol{p}}_{z} \hat{p}_{z}\right)+\boldsymbol{V}(x, y, z) . \\
\hat{\boldsymbol{H}}=\boldsymbol{i}^{2} \frac{\hbar^{2}}{2 \boldsymbol{m}}\left(\frac{\partial^{2}}{\partial x^{2}}+\frac{\partial^{2}}{\partial y^{2}}+\frac{\partial^{2}}{\partial z^{2}}\right)+\boldsymbol{V}(x, y, z) . \\
\hat{\boldsymbol{H}}=-\frac{\hbar^{2}}{2 \boldsymbol{m}} \Delta+\boldsymbol{V}(x, y, z) .
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Heisenberg's uncertainty relation

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\hat{x}\left(\hat{p}_{x} \Psi(x)\right)=x \cdot\left(-i \hbar \frac{\partial \Psi(x)}{\partial x}\right)=-i \hbar x \frac{\partial \Psi(x)}{\partial x}
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Operators whose commutator is 0 are called compatible operators!

## Stationary solutions

The Schrödinger equation of a mass point moving in the time-independent potential field:

$$
-\frac{\hbar^{2}}{2 m} \Delta \Psi(x, y, z, t)+V(x, y, z) \cdot \Psi(x, y, z, t)=i \hbar \frac{\partial \Psi(x, y, z, t)}{\partial t}
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Let's search for the solution in the next form: $\boldsymbol{e}^{-\boldsymbol{i} \frac{E}{\hbar} \boldsymbol{t}} \boldsymbol{\psi}(\boldsymbol{x}, \boldsymbol{y}, \boldsymbol{z})$. After substitution and simplification:

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\begin{gathered}
\frac{\hbar^{2}}{2 m}\left(\Delta-\frac{2 m}{\hbar^{2}} V(x, y, z)\right) e^{-i \frac{E}{\hbar} t} \psi(x, y, z)=-e^{-i \frac{E}{\hbar} t} E \psi(x, y, z) \\
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An electron in one dimension if its motion is limited to a section of length $a$.

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Therefore:

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E_{n}=\frac{\hbar^{2} \pi^{2}}{2 m a^{2}} \boldsymbol{n}^{2}
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It is generally true that if the movement of a micro-object is limited in space, its energy can only have discrete values.
Note that the smallest energy of the system is not zero, but some positive value. Zero point energy.


