

Iron-carbon equilibrium system

Fe-C system

Depending on the form of C

Fe_3C – iron-carbide (cementite)
metastable

C – graphite
stable

Possible phases

delta iron
austenite
ferrite
iron-carbide

delta iron
austenite
ferrite
graphite

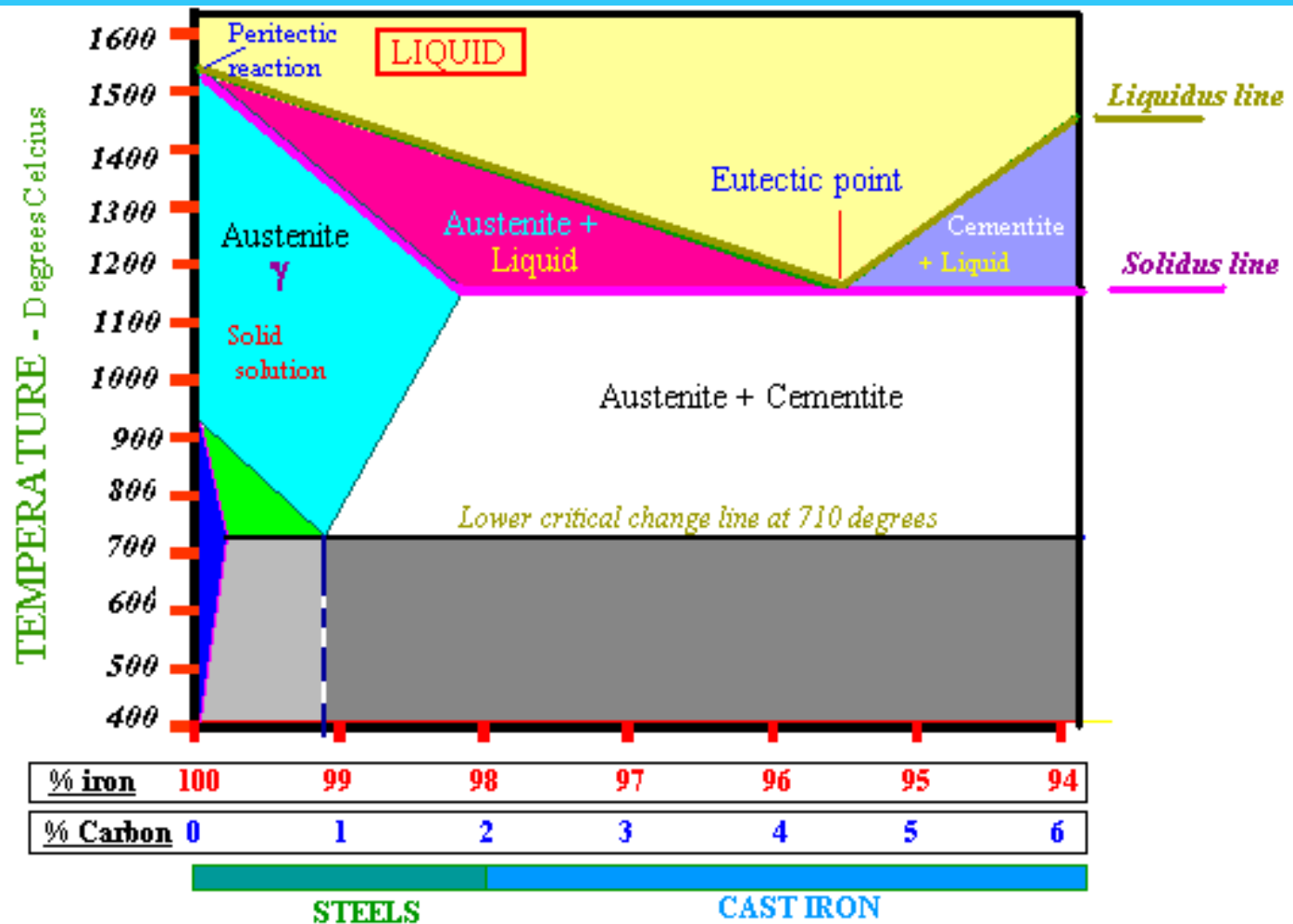
Possible structural constituents:

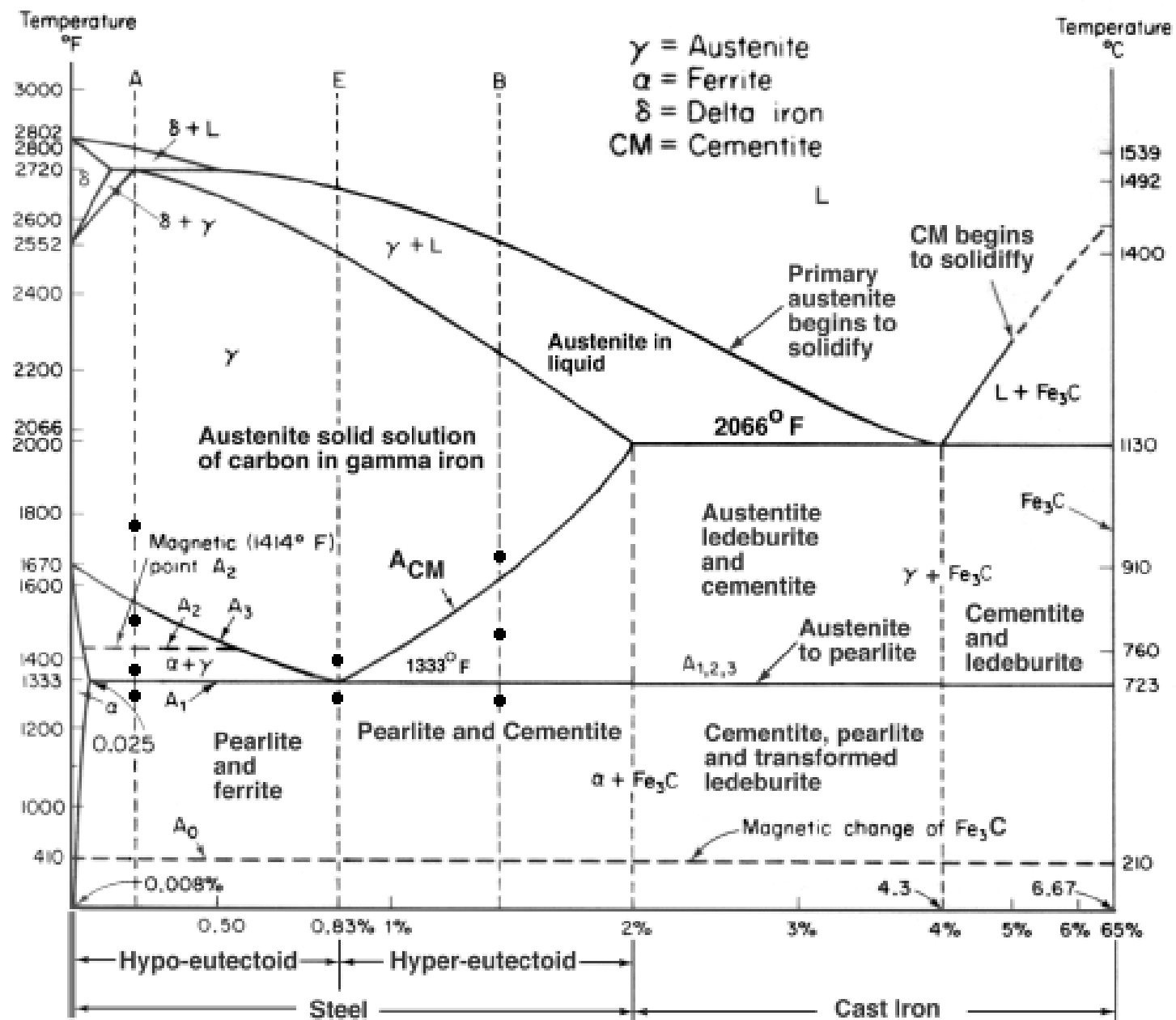
Primary, secondary, tertiary,
eutectic, eutectoid iron-carbide
austenite, ferrite,
ledeburite (eutectic)
perlite (eutectoid)

Primary, secondary, tertiary,
eutectic, eutectoid graphite
austenite, ferrite, graphitic eutectic,
and graphitic eutectoid

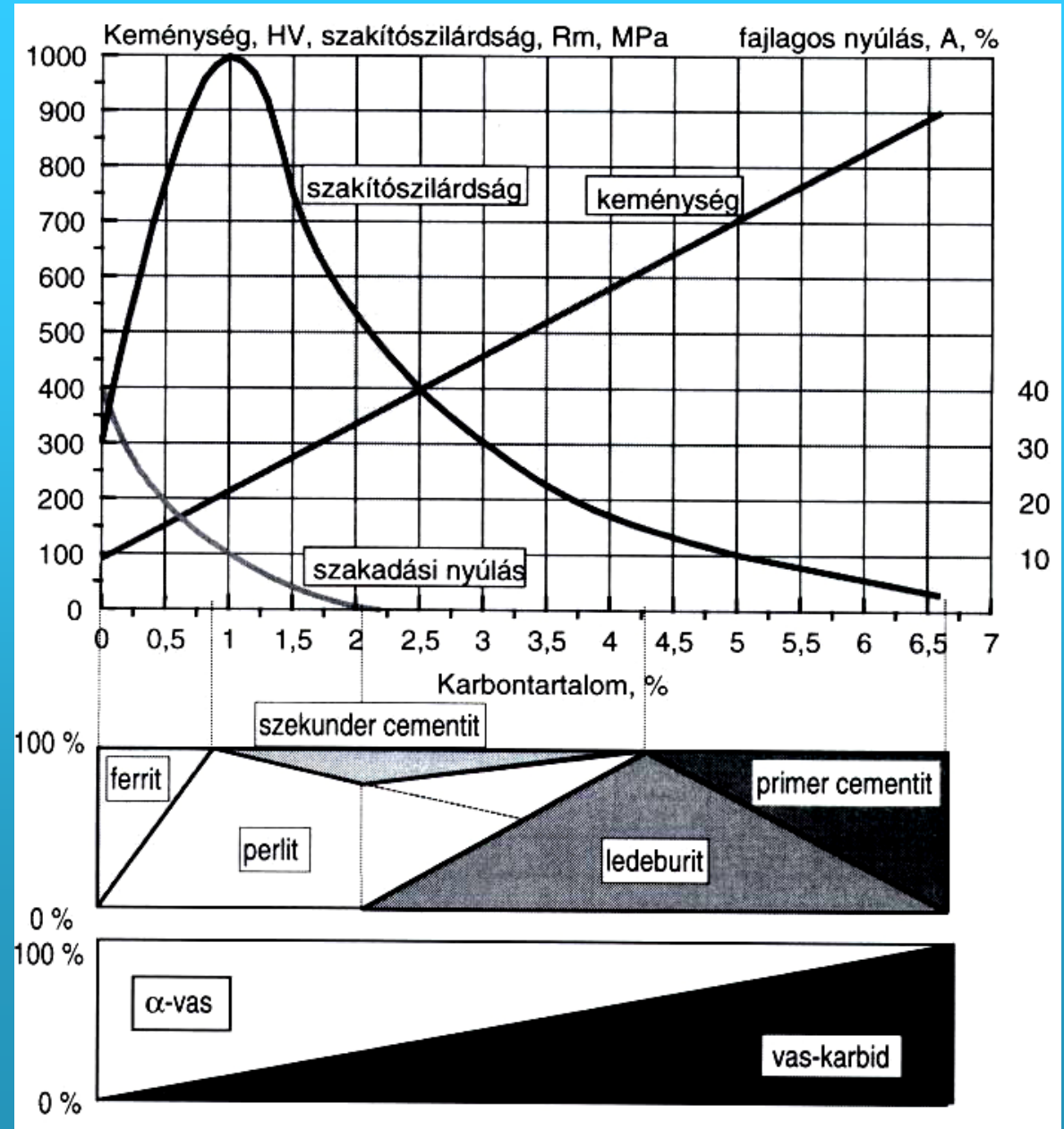
Iron-Carbon diagram

Metastable system (Fe-Fe₃C)



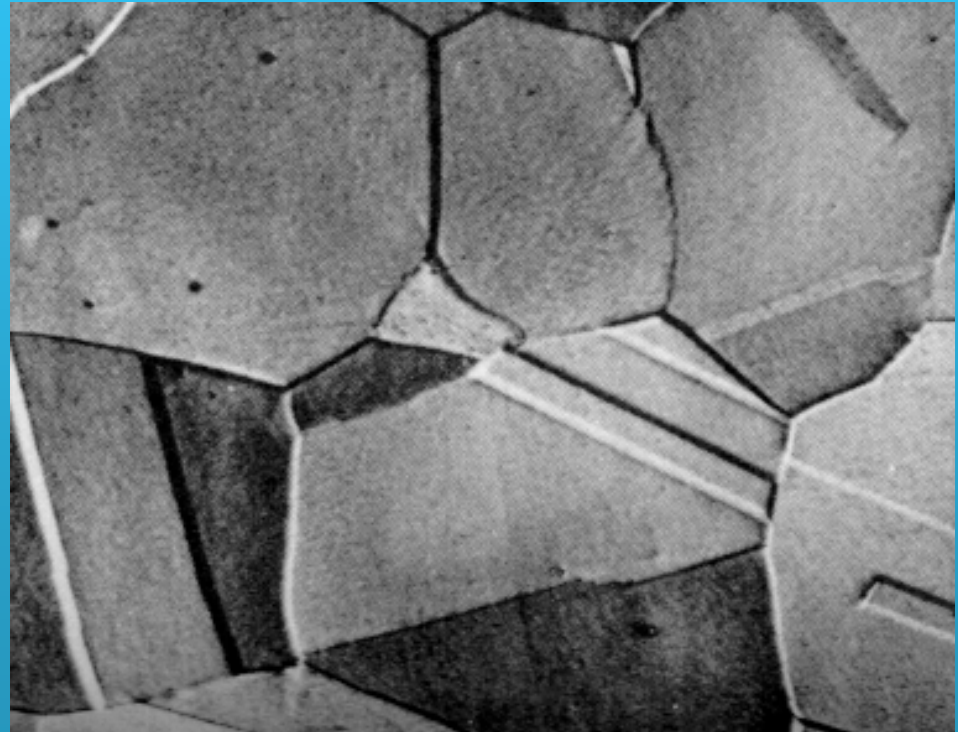


Microstructure and mechanical properties



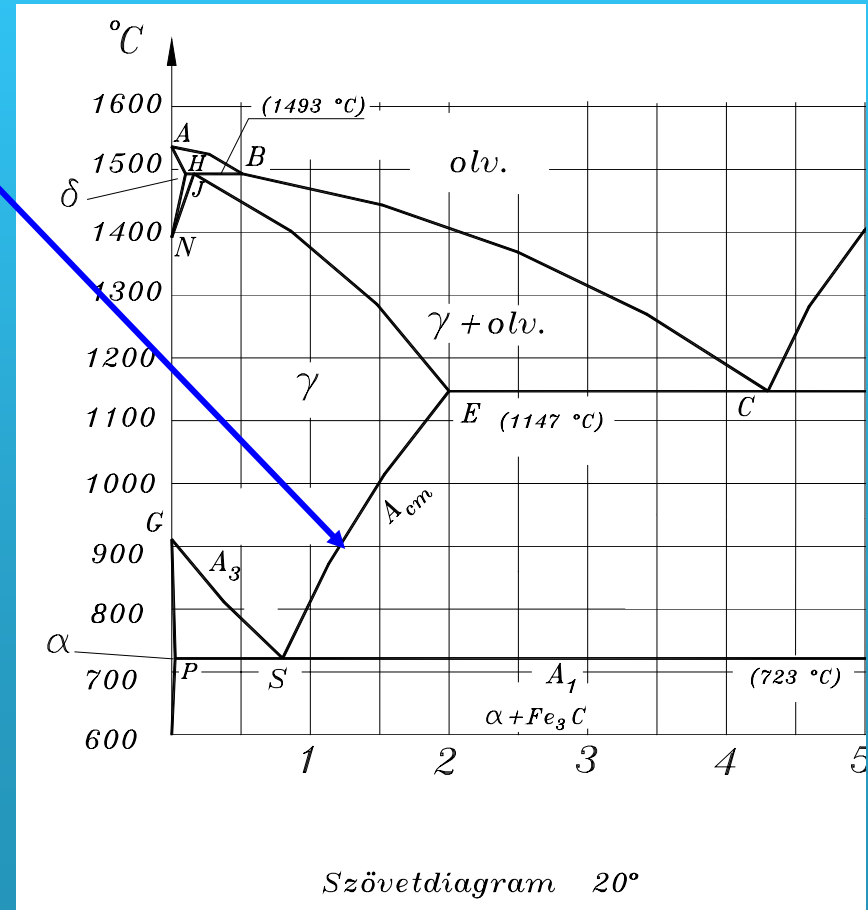
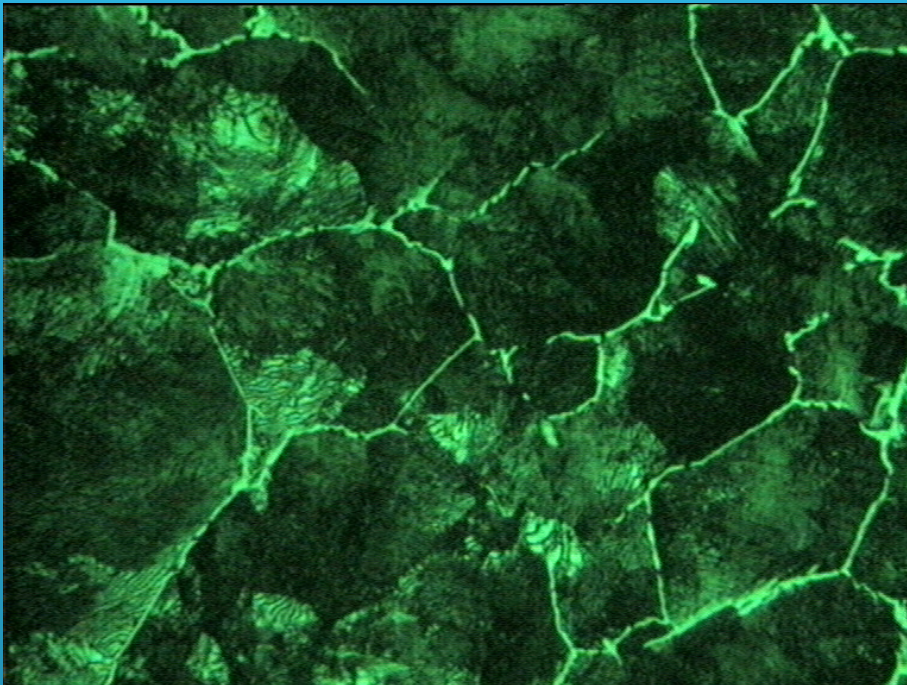
Austenite

- Interstitial solid solution
(C is solved in face centered cubic lattice of Fe)
- Limited solubility
max. solubility:
2,06% C at 1147 C°



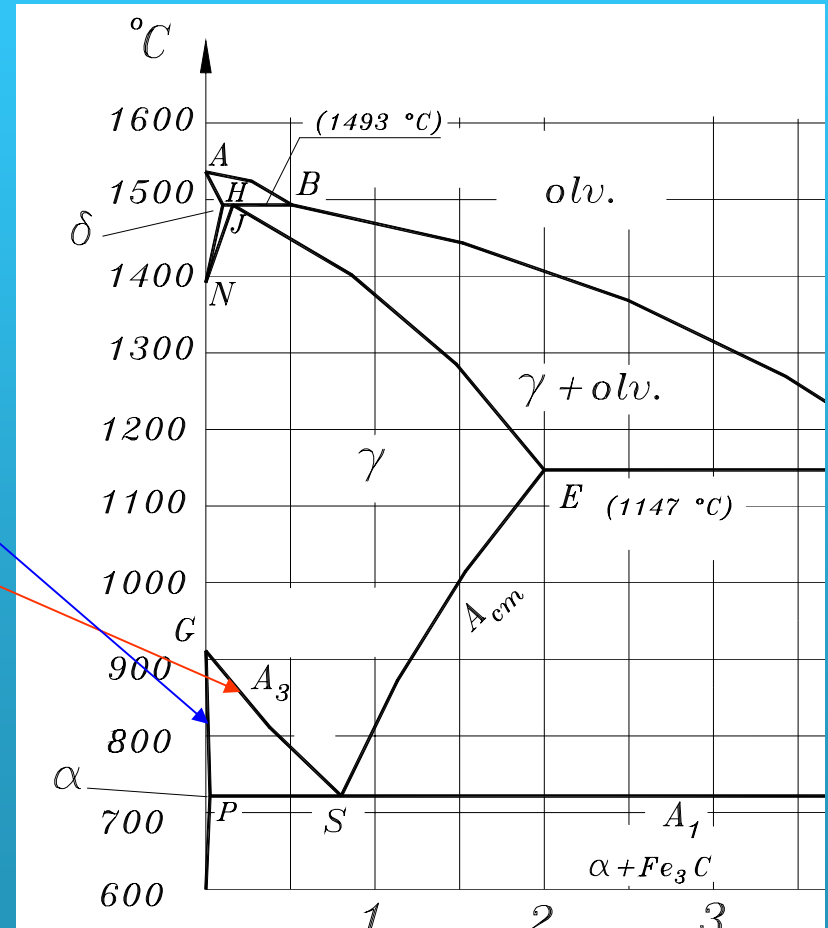
Transformations in solid state

- Limited solubility of C in austenite
- Iron-carbide segregation



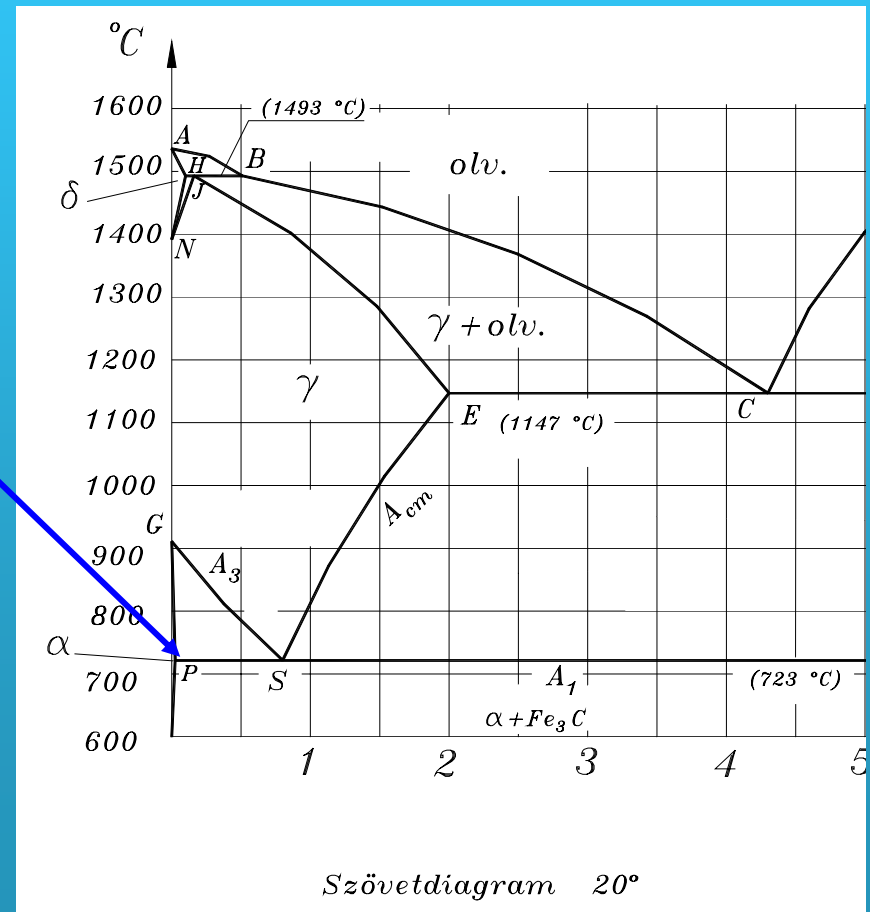
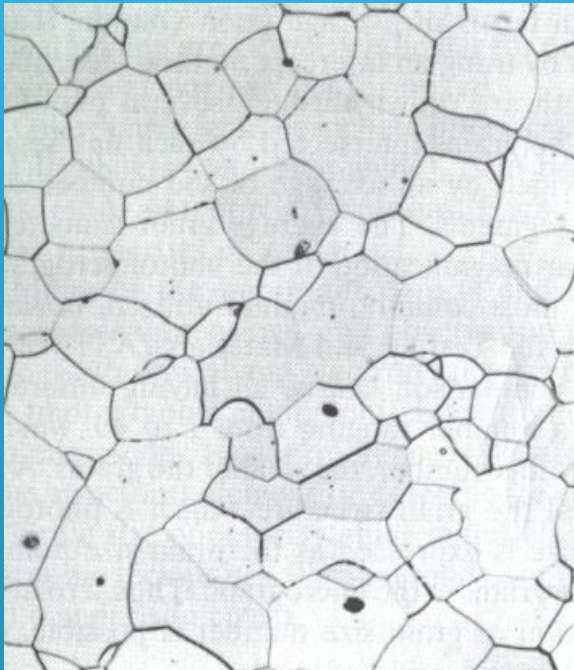
Transformations in solid state

Allotropic
transformation
of austenite into ferrite

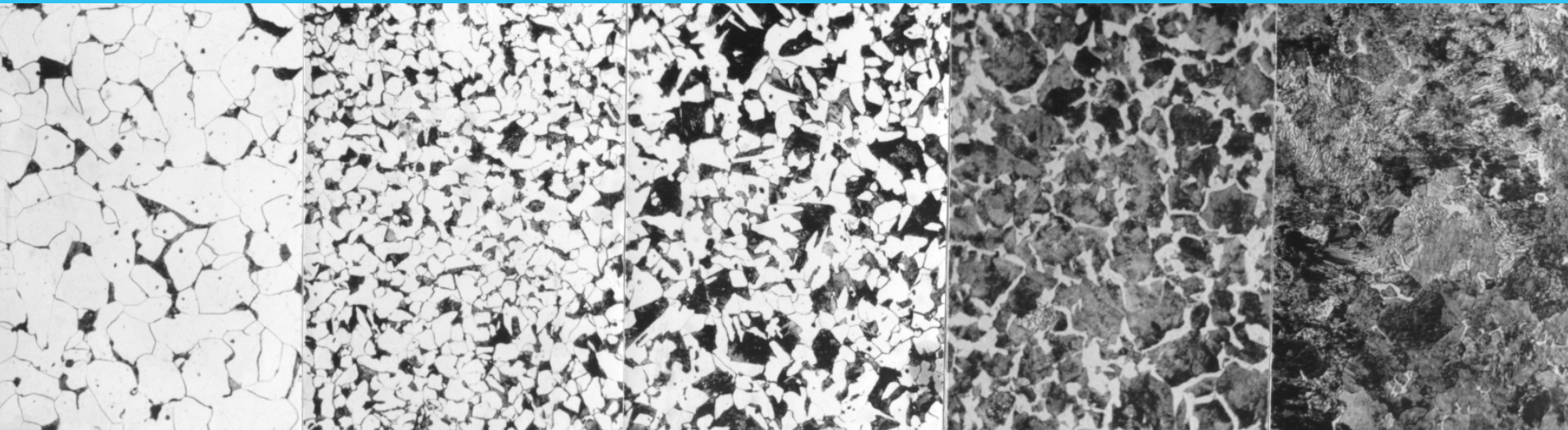


Transformations in solid state

- Limited solubility of C in ferrite
- Iron-carbide segregation



Effect of C content

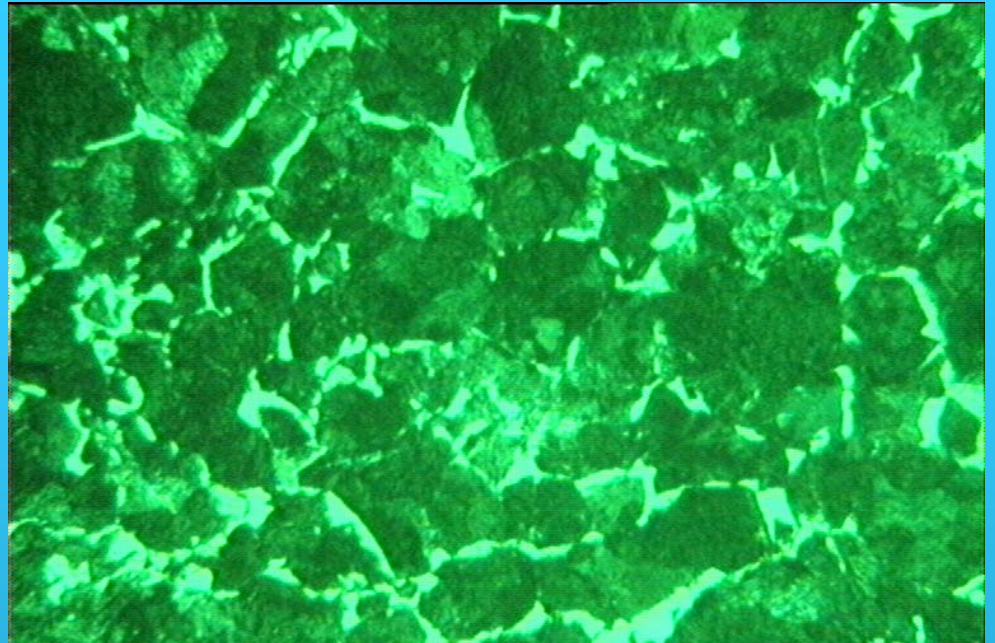


$C\% = 0,1 \dots 0,8$

Effect of C content

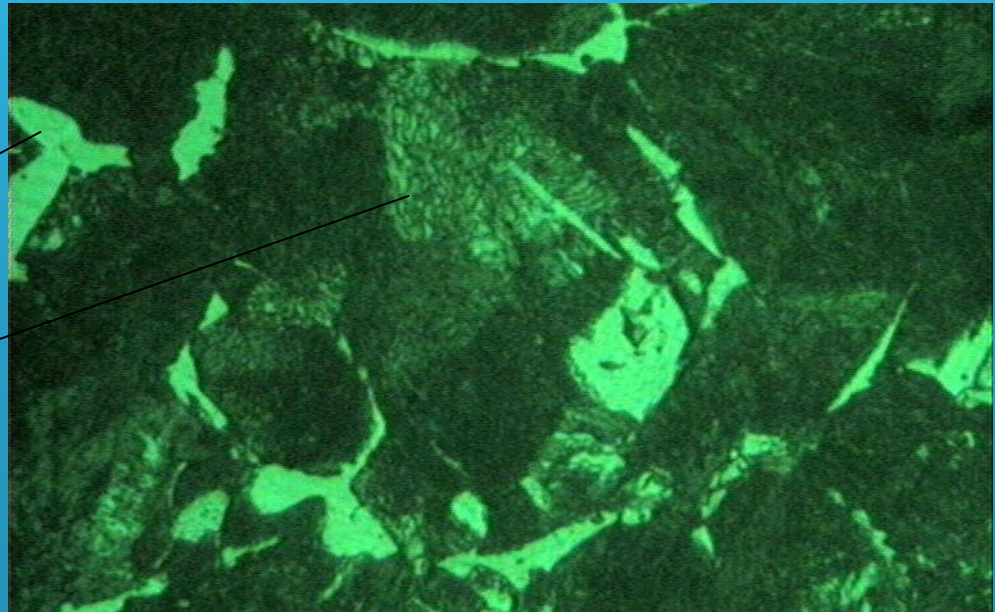
0,45 % C

Microstructure
ferrite + pearlite

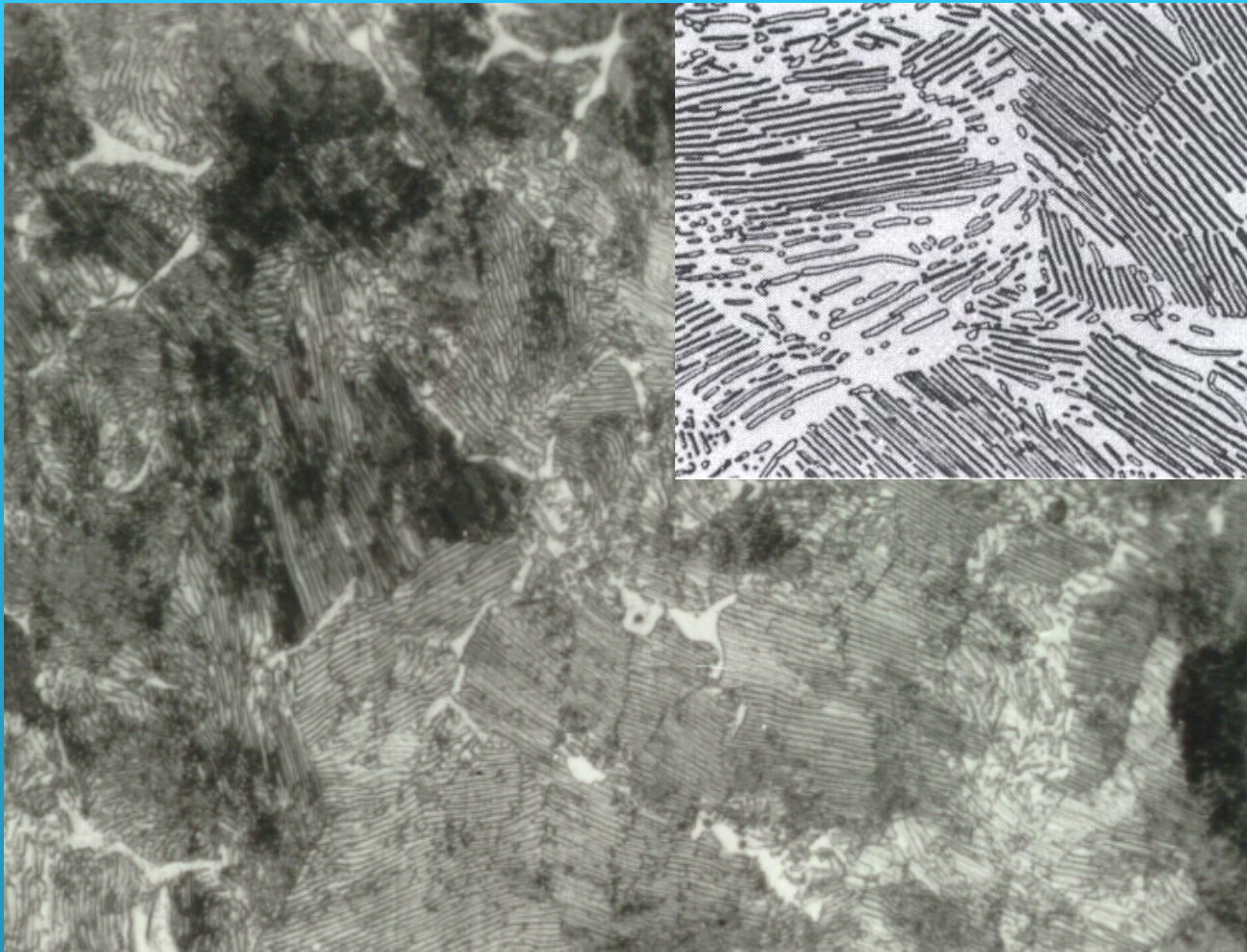


ferrite

pearlite

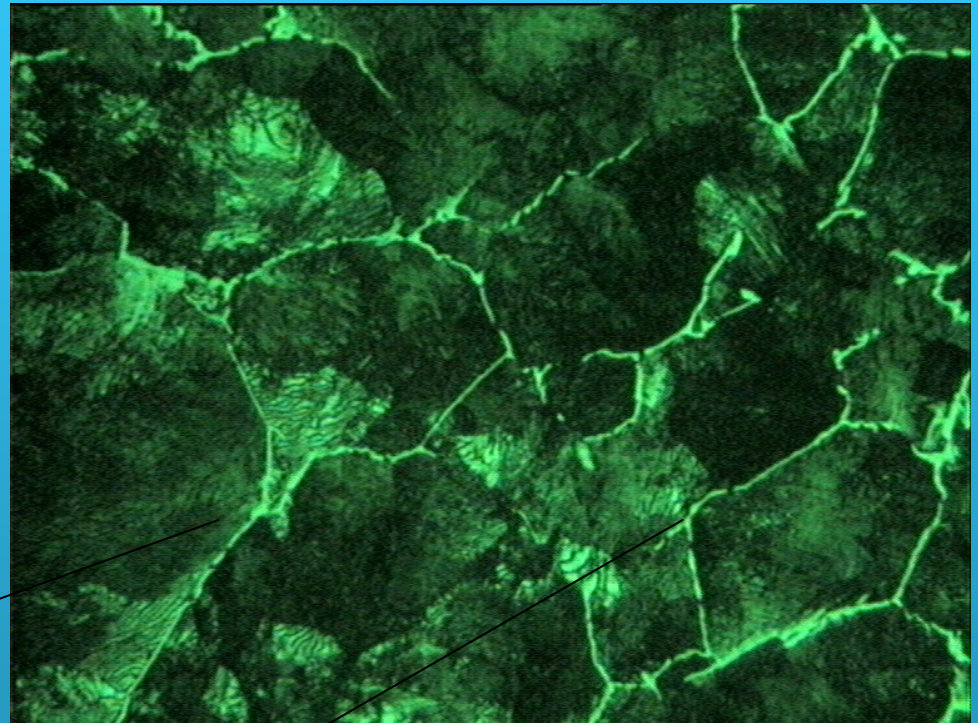


Pearlite: transformation of austenite to ferrite and cementite at 723°C; C=0,8%



Hypereutectoid steel C $\approx 1,3$ %

Microstructure
pearlite+ secondary
cementite



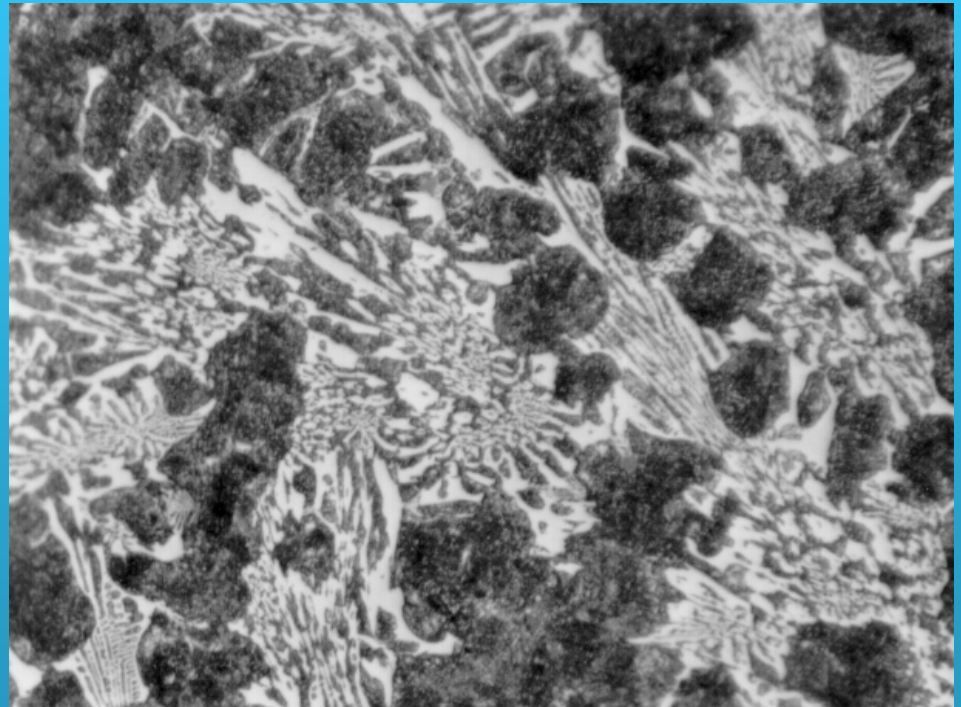
pearlite

Secondary cementite
(net)

Ledeburite (eutectic)

At 1147 C°

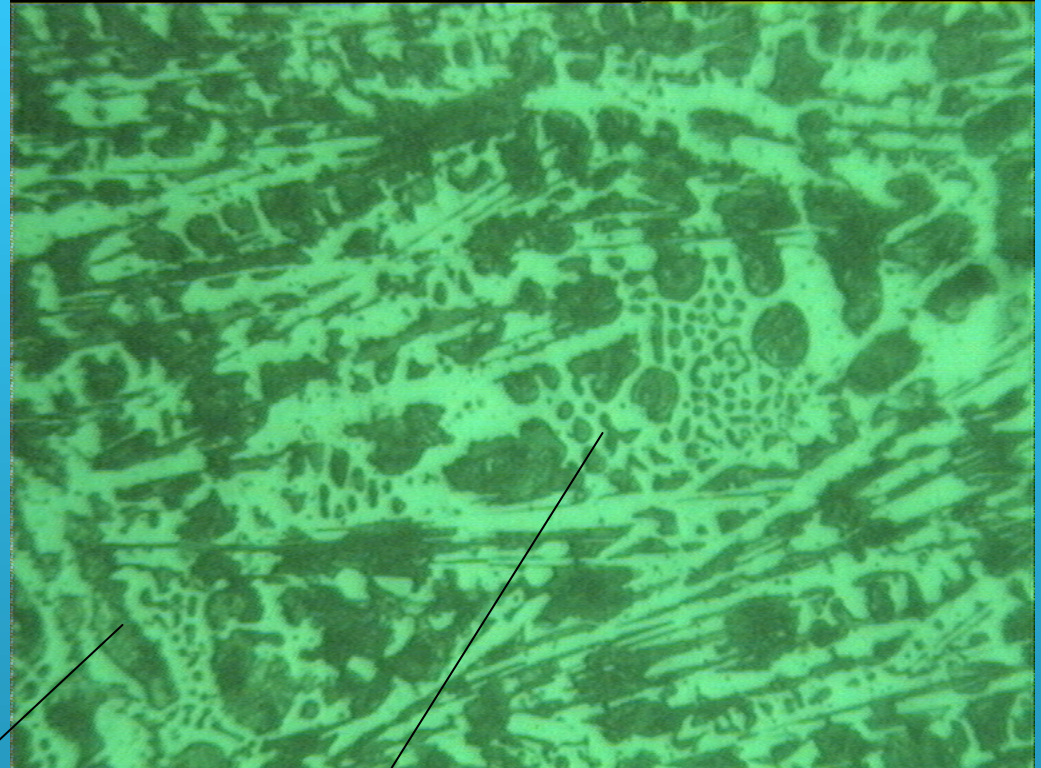
Phases of ledeburite:
austenite
iron-carbide



Ledeburite

**During cooling to room temperature:
austenite transforms to pearlite**

Hard, rigid, wear resistant



Pearlite formed
from austenite

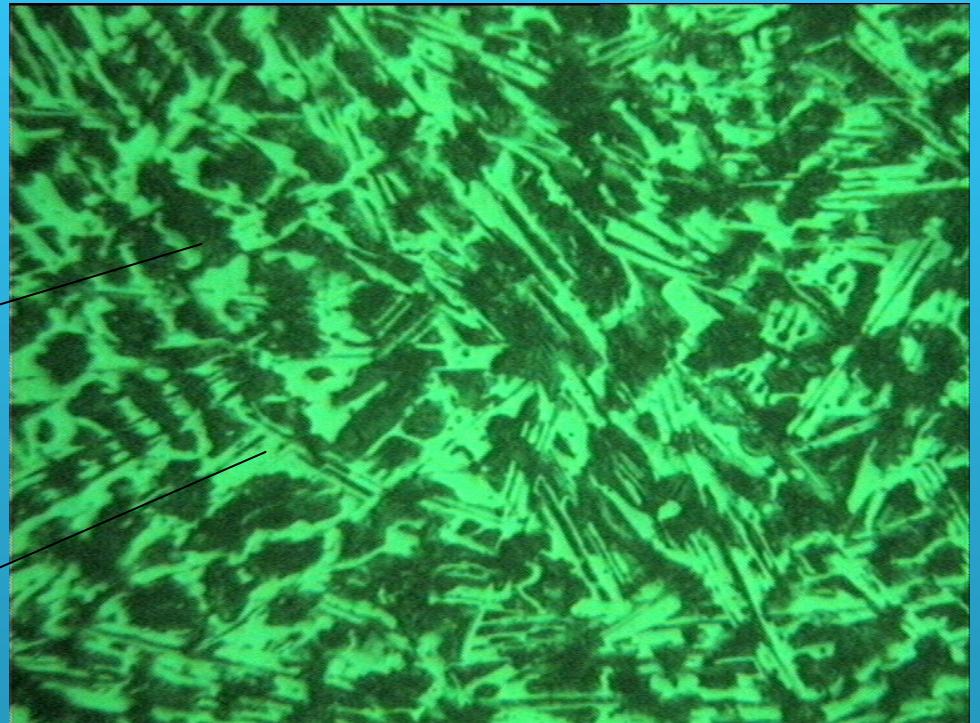
Iron-carbide

Hypoeutectic cast irons (white cast iron)

Microstructure:
**pearlite + ledeburite +
secondary cementite**

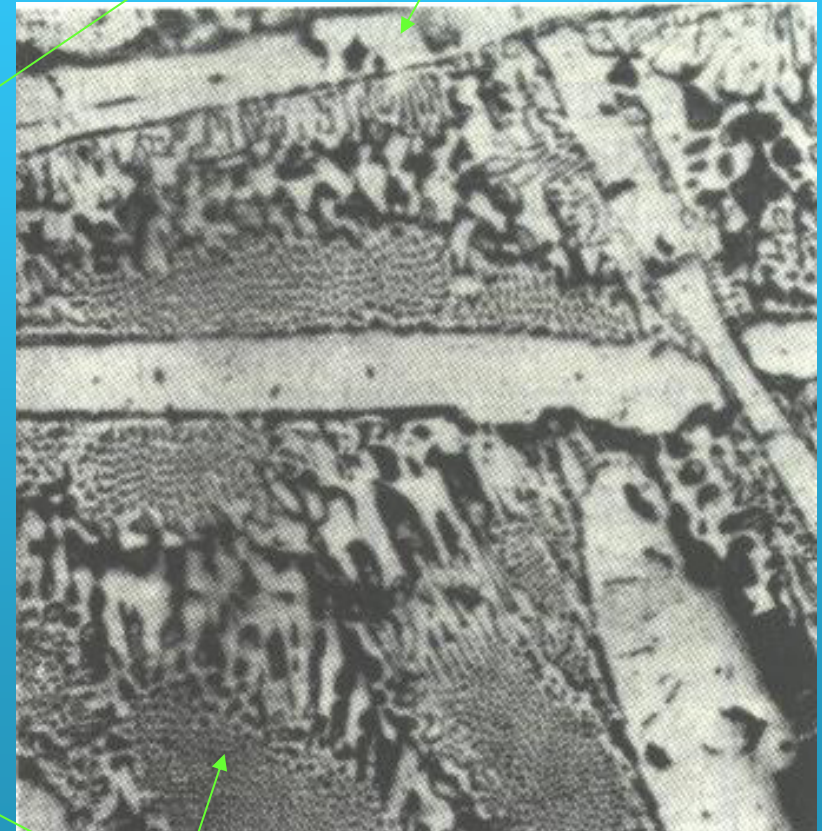
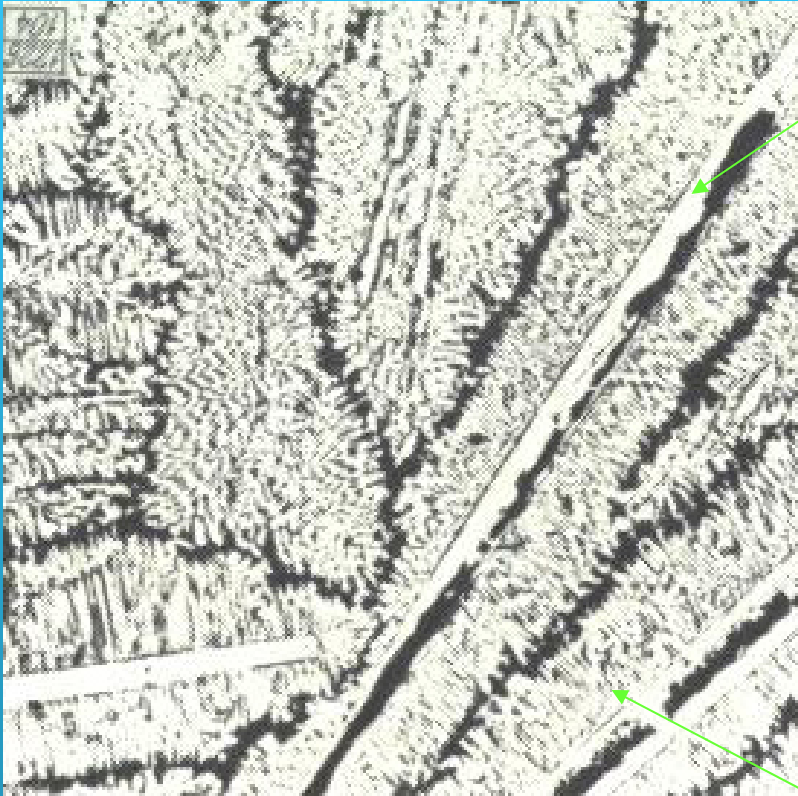
pearlite

ledeburite



Hypereutectic (white) cast irons

Primary cementite

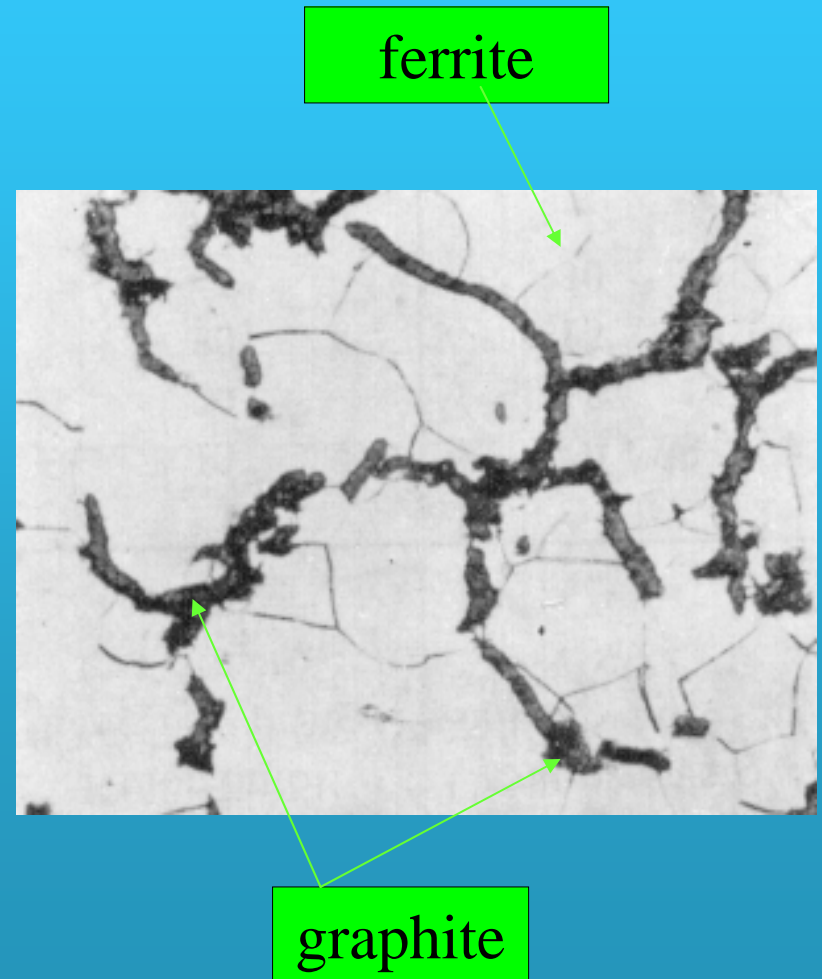


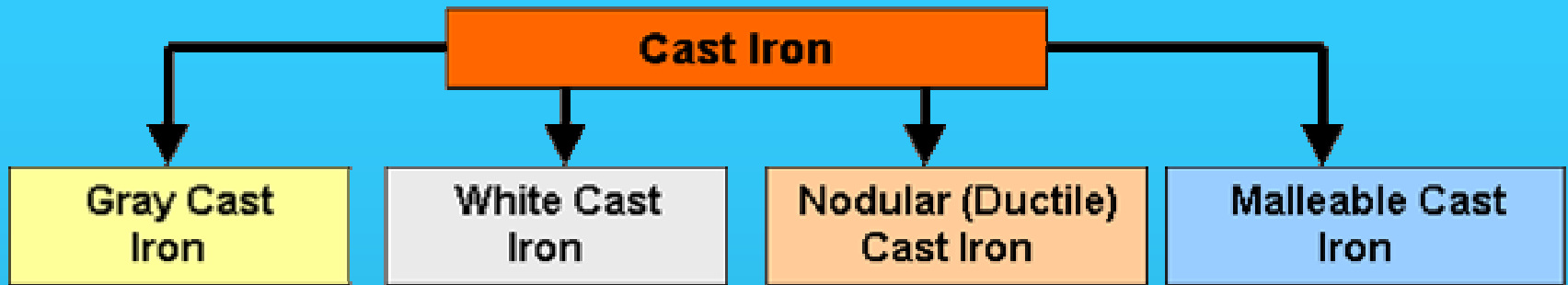
ledeburite

Hypoeutectic graphitic cast iron

- graphitic eutectic
- secondary graphite
- graphitic eutectoid

Structure at room temperature:
ferrite - graphite





Contains Si, Mn, P, S

**Effect of wall thickness of the cast part
(cooling rate)**

**Solidification and transformation
stable - metastable**

Cast irons may often be used in place of steel at considerable cost savings. The design and production advantages of cast iron include:

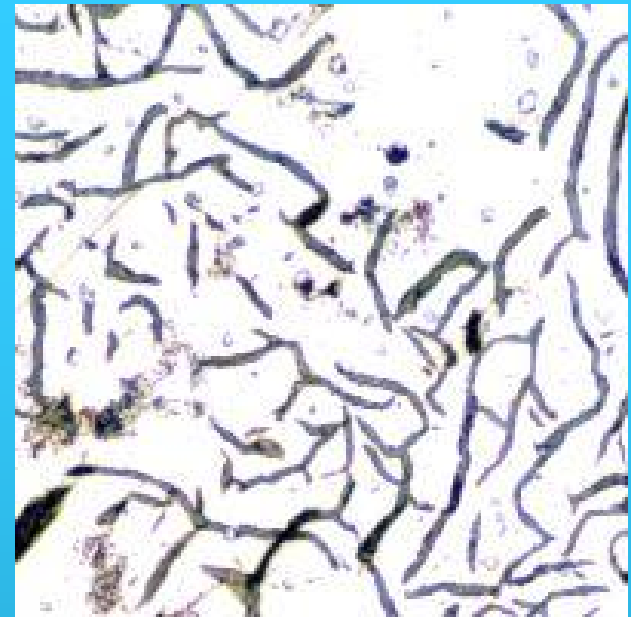
- **Low tooling and production cost**
- **Good machinability**
- **Ability to cast into complex shapes**
- **Excellent wear resistance and high hardness (particularly white cast irons)**
- **High inherent damping capabilities**

The properties of the cast iron are affected by the following factors:

- **Chemical composition of the iron**
- **Rate of cooling of the casting in the mold (which depends on the section thickness in the casting)**
- **Type of graphite formed (if any)**



**Graphite
Flakes**

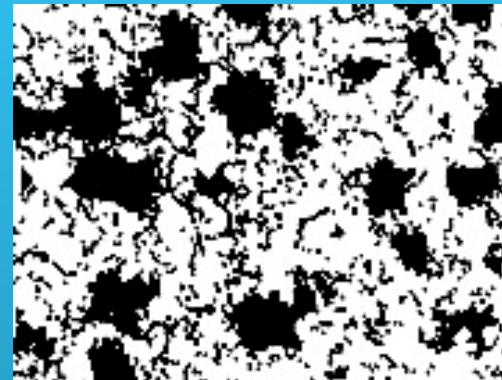
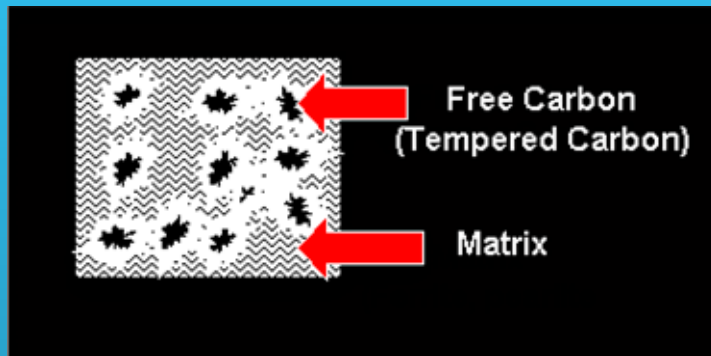
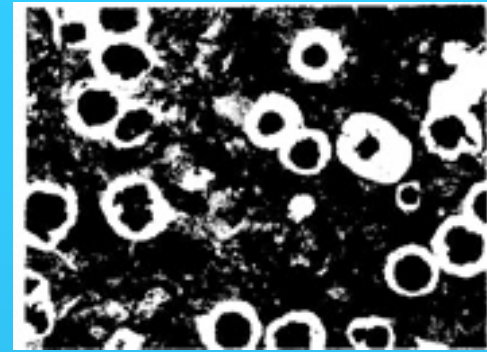
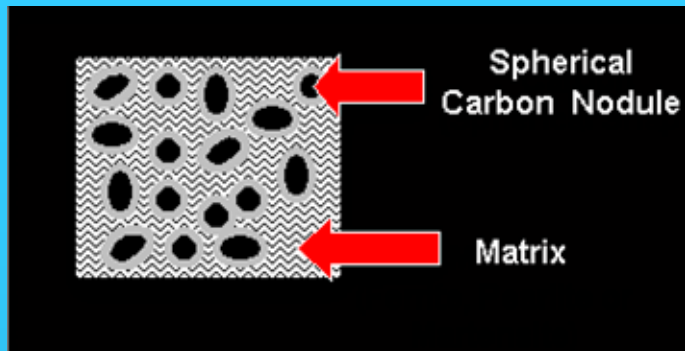


Advantages:

- Graphite acts as a chip breaker and a tool lubricant.
- Very high damping capacity.
- Good dry bearing qualities due to graphite.
- After formation of protective scales, it resists corrosion in many common engineering environments.

Disadvantages:

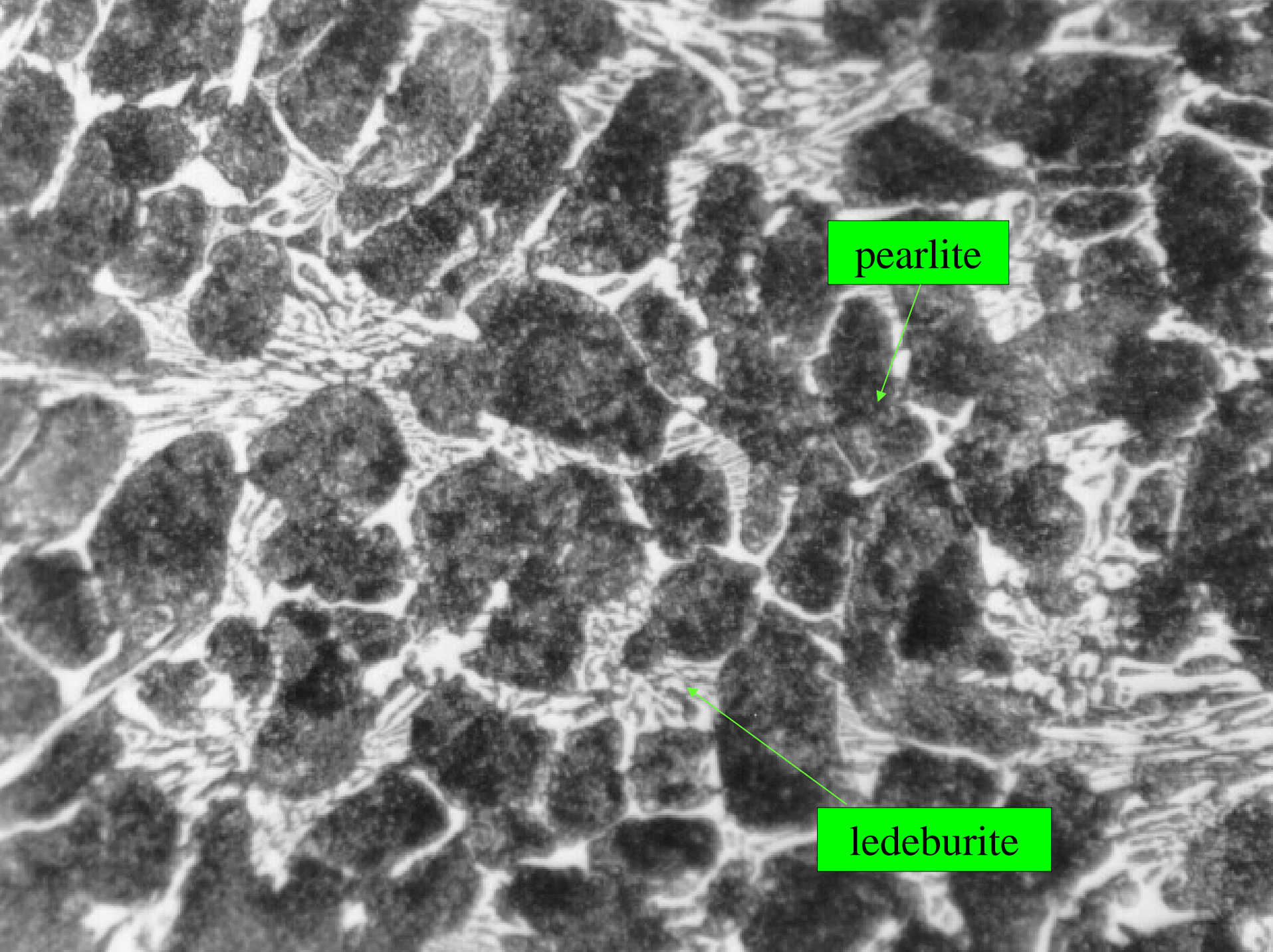
- Brittle (low impact strength) which severely limits use for critical applications.
- Graphite acts as a void and reduces strength. Maximum recommended design stress is 1/4 of the ultimate tensile strength. Maximum fatigue loading limit is 1/3 of fatigue strength.
- Changes in section size will cause variations in machining characteristics due to variation in microstructure.
- Higher strength gray cast irons are more expensive to produce.



Ductile Cast Iron

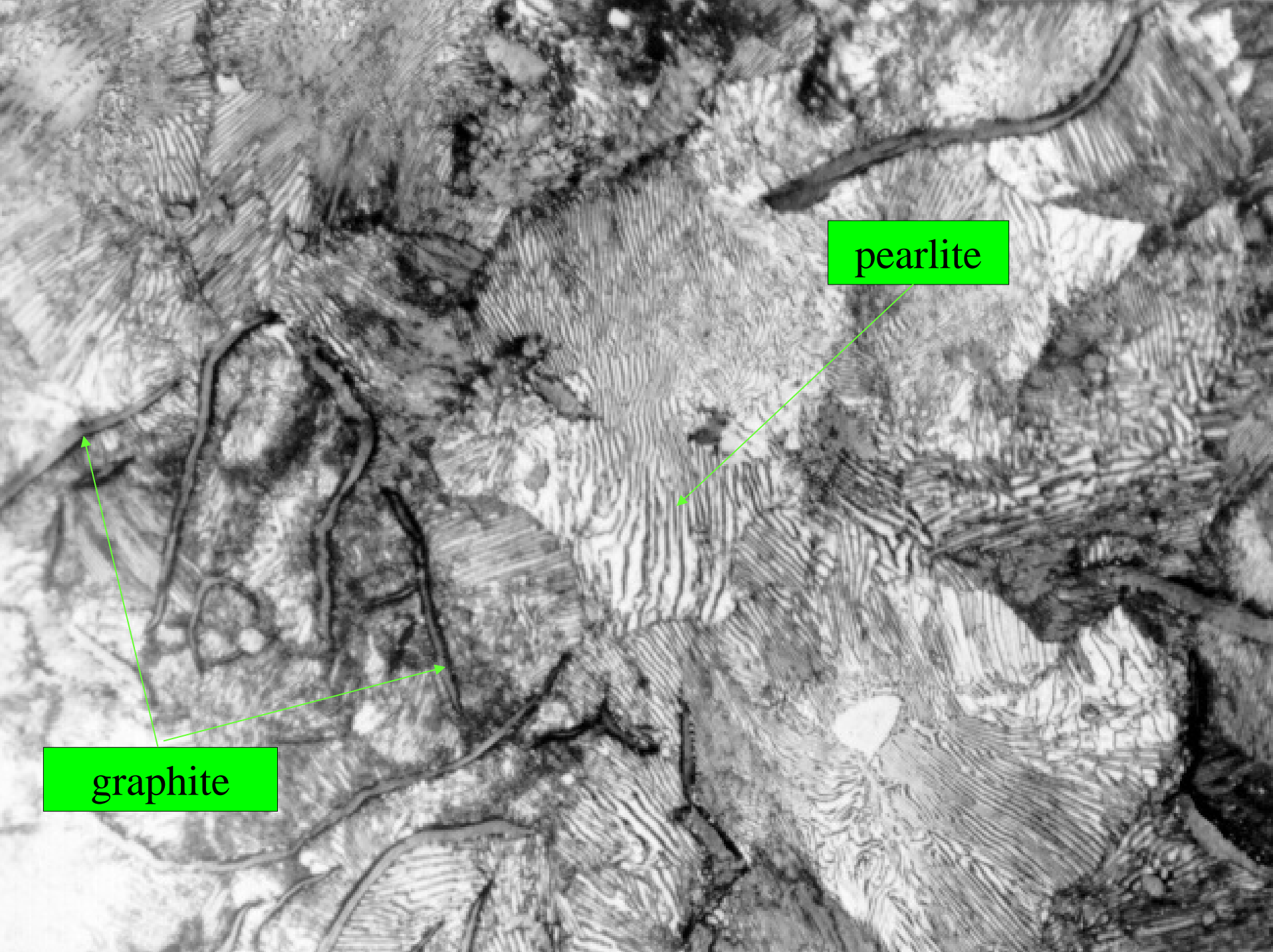
Nodular Cast Iron

Malleable Cast Iron



pearlite

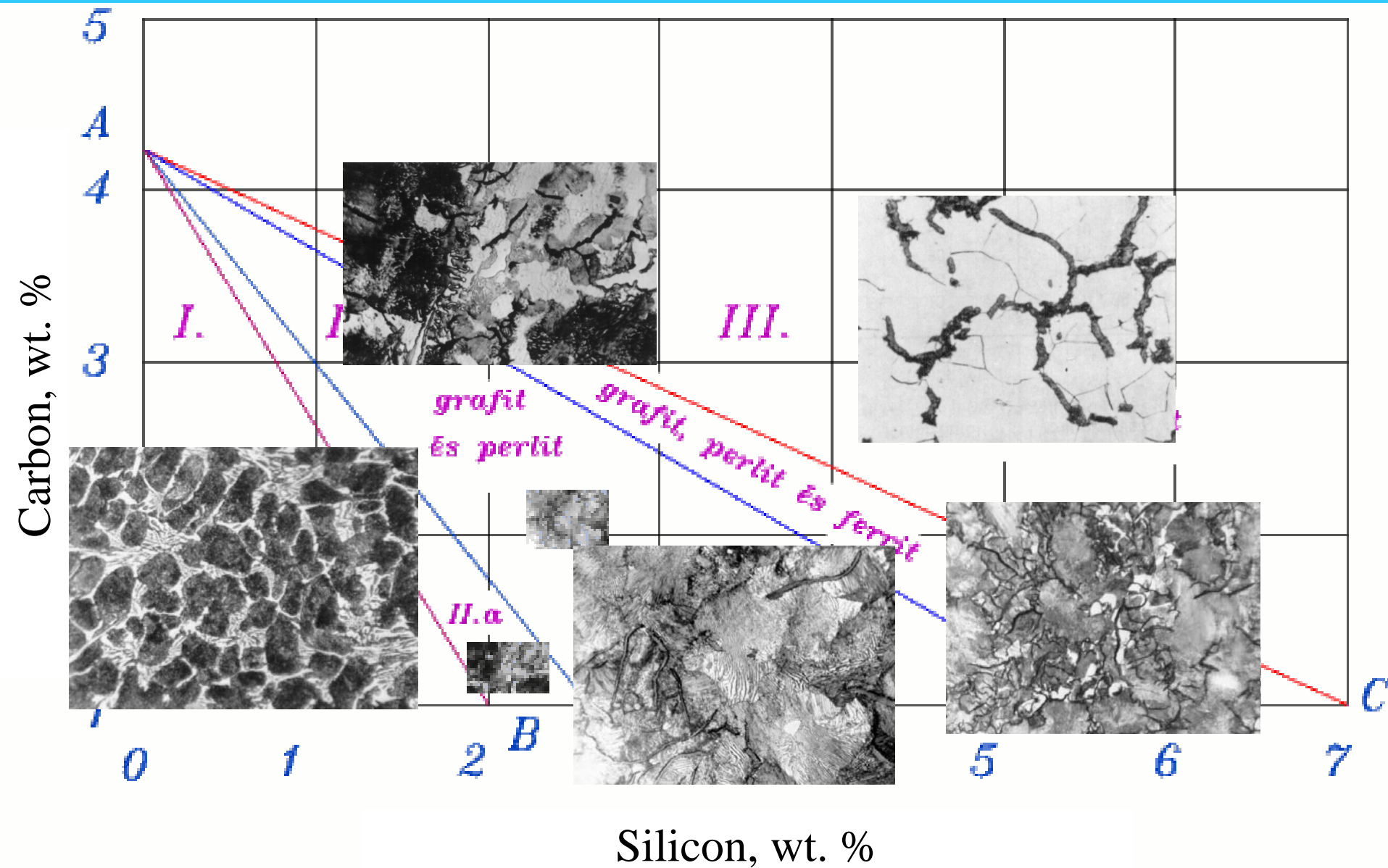
ledeburite



pearlite

graphite

Maurer diagram



Greiner - Klingenstein diagram

